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#### Key indicators

Single-crystal X-ray study T = 294 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.054 wR factor = 0.082 Data-to-parameter ratio = 19.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2002 International Union of Crystallography Printed in Great Britain – all rights reserved *N*-[3-Acetamido-1,3-bis(4-ethylphenyl)butenyl]-acetamide

Both intra- and intermolecular N-H···O hydrogen bonds are found in the structure of the title compound, C<sub>24</sub>H<sub>30</sub>N<sub>2</sub>O<sub>2</sub>. Molecules are assembled into chains, along the *b* axis, *via* intermolecular interactions. Received 7 May 2002 Accepted 13 May 2002 Online 24 May 2002

# Comment

The title compound, N-[3-acetamido-1,3-bis(4-ethylphenyl)butenyl]acetamide, (I), is one of the by-products obtained during the synthesis of N-[1-(4-ethylphenyl)ethenyl]acetamide (Burk *et al.*, 1998). In this study, the structure determination of (I) was conducted in order to obtain more knowledge about enamide compounds.



In the crystal structure (Fig. 1), the C13–C14 bond distance of 1.336 (3) Å is indicative of double-bond character. The C13–C14–C17–C22, C15–N2–C14–C17 and C15–N2– C14–C13 torsion angles are 23.8 (4), 63.2 (3) and –124.7 (3)°, suggesting little conjugation in this residue, although the C3– C13–C14–C17 torsion angle is 177.0 (2)°. Atom H2A of amine-N2 forms an intramolecular hydrogen bond with O1 (Table 1). Atom H1A of amine-N1 forms an intermolecular hydrogen bond with O2A of another molecule (x, 1/2-y, z-1/2). As illustrated in Fig. 2, these interactions link the molecules in chains along the *b* axis.

# **Experimental**

The title compound was synthesized according to the literature method of Burk *et al.* (1998). A crystal suitable for X-ray analysis was slowly grown from a mixed solvent (ethyl acetate/hexane = 1:5) at room temperature. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, Varian):  $\delta$  1.18–1.22 (*m*, 6H), 1.35 (*s*, 3H), 1.63 (*s*, 3H), 2.09 (*s*, 3H), 2.59–2.63 (*m*, 4H), 5.37 (*s*, 1H), 6.02 (*s*, 1H), 7.12–7.20 (*m*, 6H), 7.35 (*d*, *J* = 8.5 Hz, 2H), 8.97 (*s*, 1H).



## Figure 1

The molecular structure of (I), showing displacement ellipsoids at the 30% probability level (Siemens, 1995).

 $D_{\rm r} = 1.113 {\rm Mg m^{-3}}$ 

Cell parameters from 1867

Mo  $K\alpha$  radiation

reflections

T = 294 (2) K

 $R_{\rm int} = 0.083$ 

 $\theta_{\rm max} = 27.5^{\circ}$ 

 $h = -20 \rightarrow 26$ 

 $k = -7 \rightarrow 8$ 

 $l = -22 \rightarrow 16$ 

Prism, colorless

 $0.28 \times 0.23 \times 0.20 \text{ mm}$ 

5141 independent reflections

1539 reflections with  $I > 2\sigma(I)$ 

 $\begin{aligned} \theta &= 1\text{--}27.5^{\circ} \\ \mu &= 0.07 \text{ mm}^{-1} \end{aligned}$ 

Crystal data

 $\begin{array}{l} C_{24}H_{30}N_2O_2\\ M_r = 378.50\\ \text{Monoclinic, } P2_1/c\\ a = 20.773 \ (4) \ \text{A}\\ b = 6.5109 \ (10) \ \text{Å}\\ c = 17.063 \ (3) \ \text{Å}\\ \beta = 101.918 \ (4)^\circ\\ V = 2258.0 \ (6) \ \text{Å}^3\\ Z = 4 \end{array}$ 

### Data collection

Bruker CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.981, T_{\max} = 0.986$ 11299 measured reflections

#### Refinement

H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.005P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} = 0.001$
$\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.18 \text{ e } \text{\AA}^{-3}$

#### Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$N1-H1A\cdots O2^i$	0.86	2.11	2.964 (3)	175
$N2-H2A\cdots O1$	0.86	2.01	2.753 (3)	144

Symmetry code: (i)  $x, \frac{1}{2} - y, z - \frac{1}{2}$ .





H atoms were included in the riding-model approximation, with  $U_{\rm iso}$  equal to  $U_{\rm eq}$  of the atom to which they were bound.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SHELXTL-NT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-NT*; software used to prepare material for publication: *SHELXTL-NT*.

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# References

Burk, M. J., Casy, G. & Johnson, N. B. (1998). J. Org. Chem. 63, 6084–6085.Sheldrick, G. M. (1990). Acta Cryst. A46, 467–473.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXTL. University of Göttingen, Germany.

Siemens (1995). SMART (Version 5.0) and SHELXTL-NT (Version 5.10). Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.